

Diaquabis(malato- $\kappa^2O^1, O^2$ )copper(II)

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## Key indicators

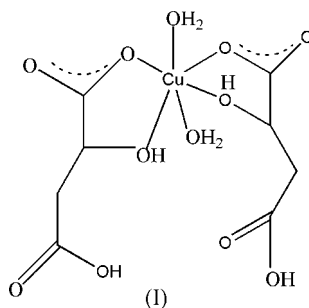
Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(C-C) = 0.002$  Å  
 $R$  factor = 0.022  
 $wR$  factor = 0.060  
Data-to-parameter ratio = 12.7For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

In the title compound,  $[Cu(C_6H_5O_5)_2(H_2O)_2]$ , the  $Cu^{II}$  atom, located on an inversion center, is coordinated by four O atoms from two malate ligands forming the equatorial plane and two water molecules in axial positions, resulting in an elongated octahedral geometry. The structure of this compound has been reported previously [Polyanskaya *et al.* (1982). *Zh. Strukt. Khim.* **23**, 117–118], but the quality of the refinement has been improved. The packing is governed by intermolecular O—H $\cdots$ O hydrogen bonds, forming an intricate three-dimensional network.

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## Comment

Some hydroxypolycarboxylic acids are present in fruits and living cells and they also play an important role in biological processes (Kotsakis *et al.*, 2003). Hydroxypolycarboxylic acids can act not only as hydrogen-bond acceptors, but also as hydrogen-bond donors, depending on the number of deprotonated carboxyl groups. The structure of the title compound, (I), has been reported previously (Polyanskaya *et al.*, 1982), but the results of this new study are of higher precision.



The  $Cu^{II}$  atom, located on an inversion center, is coordinated by four O atoms from two malate ligands which form the equatorial plane of a distorted octahedron; two water molecules occupy the axial positions with a rather long Cu—O1W distance of 2.515 (1) Å (Fig. 1). Intermolecular O—H $\cdots$ O hydrogen bonds (Table 2) result in an intricate three-dimensional network.

## Experimental

Malic acid (0.05 g, 0.82 mmol) and  $Cu(CH_3COO)_2$  (0.18 g, 1.23 mmol) were added to a mixed solvent (15 ml) of ethanol and acetonitrile (2:1) and the resulting mixture was heated for 5 h under reflux. During the process, stirring and refluxing were required. The resulting solution into which diethyl ether was allowed to diffuse in a closed vessel. A week later some single crystals of a size suitable for X-ray diffraction analysis were obtained.

## Crystal data

[Cu(C<sub>6</sub>H<sub>5</sub>O<sub>5</sub>)<sub>2</sub>(H<sub>2</sub>O)<sub>2</sub>] $M_r = 365.73$ Monoclinic,  $P2_1/c$  $a = 8.4762$  (5) Å $b = 7.4377$  (4) Å $c = 10.3117$  (6) Å $\beta = 10.268$  (10)° $V = 634.23$  (6) Å<sup>3</sup> $Z = 2$ Mo  $K\alpha$  radiation $\mu = 1.79$  mm<sup>-1</sup> $T = 293$  (2) K

0.25 × 0.18 × 0.15 mm

## Data collection

Bruker SMART CCD area-detector diffractometer

3281 measured reflections

1243 independent reflections

Absorption correction: multi-scan (SADABS; Sheldrick, 1996)

1075 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.016$  $T_{\text{min}} = 0.663$ ,  $T_{\text{max}} = 0.775$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.022$  $wR(F^2) = 0.060$  $S = 1.04$ 

1243 reflections

98 parameters

4 restraints

H-atom parameters constrained

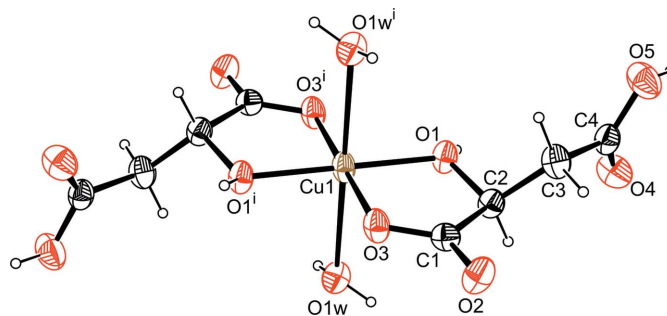
 $\Delta\rho_{\text{max}} = 0.27$  e Å<sup>-3</sup> $\Delta\rho_{\text{min}} = -0.21$  e Å<sup>-3</sup>

Figure 1

The molecular structure of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 50% probability level and H atoms are represented as small spheres of arbitrary radius. [Symmetry code: (i)  $1 - x, 1 - y, 1 - z$ .]

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINTE* (Bruker, 1999); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *SHELXTL* (Bruker, 1998).

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O1-H1\cdots O2^i$	0.83	1.80	2.6254 (17)	171
$O5-H5\cdots O1W^{ii}$	0.82	1.88	2.6874 (19)	171
$O1W-H1W\cdots O2^{iii}$	0.83	2.02	2.8353 (19)	166
$O1W-H2W\cdots O4^{iv}$	0.83	2.04	2.842 (2)	161

Symmetry codes: (i)  $x, -y + \frac{3}{2}, z + \frac{1}{2}$ ; (ii)  $x + 1, -y + \frac{3}{2}, z + \frac{1}{2}$ ; (iii)  $-x + 1, -y + 2, -z + 1$ ; (iv)  $-x + 1, y - \frac{1}{2}, -z + \frac{3}{2}$ .

All H atoms attached to C and hydroxyl O atoms were positioned geometrically and treated as riding, with C—H = 0.98 (methine) or 0.97 Å (methylene) and O—H = 0.82 Å, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  and  $1.5U_{\text{eq}}(\text{O})$ . Carboxylic H atoms were located in difference Fourier maps, but they were treated as riding on their parent O atoms, with O—H = 0.82–0.83 Å and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$ .

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## References

- Bruker (1998). *SMART* (Version 5.0) and *SHELXTL* (Version 5.1). Bruker AXS Inc, Madison, Wisconsin, USA.
- Bruker (1999). *SAINTE*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Kotsakis, N., Raptopoulou, C. P., Tangoulis, V., Terzis, A., Giapintzakis, J. & Salifoglou, A. (2003). *Inorg. Chem.* **42**, 22–31.
- Polyanskaya, T. M., Bakakin, V. V. & Zviedre, I. I. (1982). *Zh. Strukt. Khim.* **23**, 117–118.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.